

Published on Web 08/20/2009

Inhibitory Substrate Binding Site of Human Indoleamine 2,3-Dioxygenase

Changyuan Lu, Yu Lin, and Syun-Ru Yeh*

Department of Physiology and Biophysics, Albert Einstein College of Medicine, 1300 Morris Park Avenue, Bronx, New York 10461

Received April 14, 2009; E-mail: syeh@aecom.yu.edu

Indoleamine 2,3-dioxygenase (IDO) is a nonhepatic intracellular heme-containing enzyme.1 It catalyzes the conversion of Ltryptophan (L-Trp) to N-formylkynurenine (NFK), the initial and rate-determining step of the kynurenine pathway, by inserting both atoms of dioxygen across the $C_2 = C_3$ bond of the indole moiety of L-Trp.² IDO was first discovered by Hayaishi et al. more than four decades ago.³ Since then, its structural and functional properties were extensively studied until the early 1990s. This field of research was invigorated recently due to the discoveries that IDO is linked to a variety of immune-related pathophysiological conditions and that IDO is a potential target for pharmacological intervention against cancer.⁴ Here we report evidence supporting the presence of an inhibitory substrate binding site (S_i site) in human IDO (hIDO), which is capable of binding substrates (L-Trp and 1-methy-L-tryptophan), an effector (3-indole ethanol), and an uncompetitive inhibitor (Mitomycin C). The structures of these molecules are illustrated in Scheme 1.

Scheme 1. Molecular Structures of 3-Indole Ethanol (IDE), Mitomycin C (MtoC), and 1-Methyl-tryptophan (1MTrp)



The activity of hIDO was first examined as a function of L-Trp concentration, following the mixing of the ferric enzyme with a methylene blue-ascorbate reducing system.³ As shown in Figure 1a, the activity of hIDO follows typical Michaelis-Menten behavior at $[L-Trp] < 50 \ \mu\text{M}$; further increase in [L-Trp] caused a decrease in the activity, signifying substrate-inhibition behavior, well-known for rabbit IDO (rIDO).^{3,5} It was generally believed that, at high [L-Trp], the substrate can bind to the ferric enzyme, thereby retarding the turnover of the enzyme by inhibiting its reduction to the active ferrous state.⁵ This scenario, however, can be excluded on the basis of two new observations: (1) the dissociation constant of L-Trp for the ferric IDO ($K_d = 0.9$ mM; see Figure S1) is significantly higher than the self-inhibition constant, K_{si} (0.17 mM, vide infra), and (2) the redox potential of the L-Trp-bound ferric hIDO is \sim 46 mV higher than that of the substrate-free enzyme,⁶ indicating L-Trp binding to the ferric enzyme does not prevent its reduction.

We propose that the substrate-inhibition behavior of hIDO is a result of the binding of a second L-Trp in an inhibitory S_i site of the enzyme. This hypothesis is consistent with recent computational data showing that the distal pocket of hIDO is flexible enough to accommodate both L-Trp and an indole derivative.⁷ On the basis of this scenario, the data shown in Figure 1a were fitted with the following equation⁸ by assuming L-Trp can bind to the active site, as well as the S_i site.



Figure 1. Michaelis–Menten plots of the dioxygenase reaction of hIDO (91nM) at pH 7.4 with respect to [L-Trp] (a) and [D-Trp] (b) under airsaturated conditions. The inset in (b) shows the corresponding plot for the reaction of hIDO with 36 μ M L-Trp with respect to [O₂] at pH 7.4.

$$V = V_{\text{max}} \times [S]/(K_{\text{m}} + [S] \times (1 + [S]/K_{\text{si}}))$$
(1)

Here, [S] is the substrate concentration; V_{max} and K_{m} are Michaelis—Menten constants. The best-fitted parameters are listed in Table 1. The data indicate that, under steady-state conditions, the probability of L-Trp binding to the ferric enzyme, followed by its reduction to the active ferrous state, is negligible (similar to that concluded by Sono et al. for rIDO⁵), as the L-Trp dissociation constant of the ferric enzyme ($K_d = 0.9 \text{ mM}$) is much higher than the K_m (15 μ M) and K_{si} (0.17 mM). It is noteworthy that the physiological concentration of L-Trp, ~50–100 μ M,⁹ lies right between the observed K_m and K_{si} values.

Table 1. Michaelis-Menten and Inhibition Constants of hIDOAssociated with Its Reaction with L-Trp in the Presence andAbsence of IDE and Those Associated with D-Trp and L-1MTrp(k_{cat} Values Calculated from $V_{max}/[hIDO]$)

	∟-Trp	$+ IDE_{2.5mM}$	$+ IDE_{5.1mM}$	D-Trp	∟-1MTrp
$\frac{k_{\text{cat}} (\text{s}^{-1})}{K_{\text{m}} (\mu \text{M})}$ $K_{\text{si}} (\text{mM})$	$\begin{array}{c} 3.1 \pm 0.2 \\ 15 \pm 2 \\ 0.17 \pm 0.2 \end{array}$	$\begin{array}{c} 3.2 \pm 0.4 \\ 16 \pm 3 \\ 0.48 \pm 0.05 \end{array}$	$\begin{array}{c} 3.5 \pm 0.6 \\ 15 \pm 3 \\ 1.0 \pm 0.1 \end{array}$	$5.9 \pm 0.3 \\ (2.6 \pm 0.2) \times 10^{3} \\ -$	$\begin{array}{c} 0.064 \pm 0.003 \\ 62 \pm 9 \\ 5.0 \pm 0.8 \end{array}$

Similar activity studies were carried out with D-Trp. No substrate inhibition was observed (Figure 1b), indicating substrate inhibition is stereospecific to the L isomer, as reported for rIDO.³ The Michaelis-Menten fit of the data shows, as compared to the L-Trp reaction, the k_{cat} is 2-fold faster (5.9 s⁻¹), while the $K_{\rm m}$ value is 170-fold larger (2.6 mM). The data demonstrate that the substrate stereoselectivity of hIDO is a result of its preferential binding of the L-isomer. Additional activity studies were conducted as a function of $[O_2]$ in the presence of 36 μ M L-Trp (see the insert in Figure 1b). The k_{cat} value was determined to be 2.9 s⁻¹, similar to that determined by the data shown in Figure 1a. $K_{\rm m}$ for O₂ was determined to be 42 μ M, which is near the physiological level of oxygen, $\sim 50-76 \ \mu M$,¹⁰ and is comparable to that of an analogous enzyme, tryptophan 2,3dioxygenase, $(40 \ \mu M)^{11}$ and those of monooxygenases, such as NOSs $(6-24 \ \mu M)^{12}$ or P450s $(1-15 \ \mu M)$.¹³

IDE has been long recognized as an effector for rIDO.¹⁴ It was believed IDE enhances the activity of rIDO by improving its k_{cat}

via binding to an accessory binding site located near the catalytic site of IDO.¹⁴ To investigate if the IDE binding site coincides with the aforementioned S_i site, we examined the hIDO activity in the presence of 2.5 and 5.1 mM IDE. As shown in Figure 2a, the presence of IDE led to higher activity and less pronounced substrate inhibition. The best fit of data with eq 1 indicates IDE does not affect k_{cat} and K_m but causes the elevation of K_{si} from 0.17 to 0.48/ 1.0 mM (for 2.5/5.1 mM IDE, see Table 1), indicating IDE competes with L-Trp for the S_i site. As IDE bound to the S_i site, unlike L-Trp, does not retard the active site catalysis, the apparent activity is higher than that in its absence. The data clearly demonstrate that IDE acts as an effector by binding to the S_i site, thereby blocking it from L-Trp binding, instead of by facilitating k_{cat} , as generally believed.

MtoC, a quinone-containing natural product from Streptomyces, has been widely used as a chemotherapy drug for several types of cancer.¹⁵ It is believed that the antitumor activity of MtoC is a result of its ability to generate oxygen radicals upon reduction, thereby inhibiting DNA synthesis. The recent findings that hIDO is a potential therapeutic target for cancer treatment and that quinone is a potent pharmacophore for hIDO inhibition¹⁶⁻¹⁸ prompted us to investigate if hIDO could function as a pharmacological target of MtoC. We examined the hIDO activity as a function of MtoC concentration. The Lineweaver-Burk plot of the data (Figure 2b) shows MtoC indeed inhibits hIDO in an uncompetitive fashion with an inhibition constant (K_i) of ~25 μ M. The data indicate that, as an uncompetitive inhibitor, MtoC binds only to the L-Trp-bound enzyme, not the substrate-free enzyme. We propose that L-Trp binding in the active site induces structural changes to the S_i site to accommodate the MtoC molecule and that the occupancy of the S_i site by MtoC inhibits the active site catalysis.

As a control experiment, the inhibitory effect of 1MTrp was examined. 1MTrp has been widely used as an hIDO inhibitor.¹⁹ As shown in Figure 2c, the L-isomer acts as a competitive inhibitor, with a K_i of 32 μ M, while the D-isomer exhibits no inhibitory effect (data not shown), consistent with that reported by Hou et al.²⁰ Additional studies show, in the absence of L-Trp, L-1MTrp itself can act as a substrate, although k_{cat} is 50-fold lower than that of L-Trp (Figure S2), similar to that reported by Chauhan et al.²¹ In addition, our data show that L-1MTrp exhibits substrate-inhibition behavior, indicating, like L-Trp, L-1MTrp binds to the S_i site at elevated concentrations, thereby inhibiting hIDO activity. The 29fold higher K_{si} value, as compared to that of L-Trp (Table 1), indicates the bulky methyl group on the indole nitrogen significantly lowers its affinity toward the S_i site.

In summary, our data demonstrate that hIDO possesses two substrate binding sites, an active binding site and a S_i site (as illustrated in a cartoon shown in Figure S3). The binding of L-Trp in the substrate binding site introduces a conformational change to the S_i site to allow it to accommodate molecules with a wide variety of structures, including L-Trp, 1MTrp, IDE, and MtoC. We demonstrated that, by binding to the S_i site, these molecules can act as either an effector or an inhibitor. It is important to note that L-Trp-binding induced conformational changes to hIDO have been implicated in the crystallographic data,²² as well as molecular dynamics simulations,⁷ while the allosteric structural transition induced by the binding of L-Trp, L-1MTrp, or MtoC in the S_i site, which gives rise to the inhibition of the active catalysis, remains to be further investigated. Nonetheless, the data presented in this work offer new mechanistic insights into the substrate-inhibition behavior of hIDO, as well as the functional mechanisms of its effector, IDE, and inhibitor, MtoC.



Figure 2. Michaelis-Menten plots of the hIDO reaction at pH 7.4, in the absence or presence 2.5 or 5.1 mM IDE (a) and the Lineweaver-Burk plots of the steady-state activities of hIDO in the presence of various concentrations of MtoC (b) and L-1MTrp (c).

Recently, hIDO has emerged as a therapeutic target for cancer,⁴ leading to an active search for potent inhibitors. Our data show MtoC effectively inhibits hIDO, calling for re-evaluation of the action mechanism of this commonly used antitumor chemotherapeutic agent. They also suggest the newly discovered quinone-containing hIDO inhibitors,^{16–18} like MtoC, may selectively bind to the S_i site. Taken together the data presented in this work provide the first glimpse of the S_i site that offers potential guidelines for future development of more efficient hIDO inhibitors.

Acknowledgment. We would like to thank Dr. Denis L. Rousseau for valuable discussions.

Supporting Information Available: The Materials and Methods, the absorption spectra of the substrate-free and L-Trp-bound ferric hIDO and the associated L-Trp titration data, as well as the Michaelis—Menten plot with respect to L-1MTrp and a cartoon illustrating the S_i site of hIDO. This material is available free of charge via the Internet at http://pubs.acs.org.

References

- (1) Takikawa, O. Biochem. Biophys. Res. Commun. 2005, 338, 12-19.
- (2) Sono, M.; Roach, M. P.; Coulter, E. D.; Dawson, J. H. Chem. Rev. 1996, 96 (7), 2841–2888.
- (3) Yamamoto, S.; Hayaishi, O. J. Biol. Chem. 1967, 242 (22), 5260-6.
- (4) Muller, A. J.; DuHadaway, J. B.; Donover, P. S.; Sutanto-Ward, E.; Prendergast, G. C. Nat. Med. 2005, 11 (3), 312–9.
- Sono, M.; Taniguchi, T.; Watanabe, Y.; Hayaishi, O. J. Biol. Chem. 1980, 255 (4), 1339-45.
- (6) Papadopoulou, N. D.; Mewies, M.; McLean, K. J.; Seward, H. E.; Svistunenko, D. A.; Munro, A. W.; Raven, E. L. *Biochemistry* 2005, 44 (43), 14318–28.
- (7) Macchiarulo, A.; Nuti, R.; Bellocchi, D.; Camaioni, E.; Pellicciari, R. Biochim. Biophys. Acta 2007, 1774 (8), 1058–68.
- 8) Copeland, R. A. Enzymes, 2nd ed.; Wiley-VCH: New York, 2000.
- (9) Torres, M. I.; Lopez-Casado, M. A.; Lorite, P.; Rios, A. Clin. Exp. Immunol. 2007, 148 (3), 419–24.
- (10) Bunn, H. F.; Poyton, R. O. Physiol. Rev. 1996, 76 (3), 839-85.
- (11) Ishimura, Y.; Nozaki, M.; Hayaishi, O. J. Biol. Chem. 1970, 245 (14), 3593–602.
- (12) Rengasamy, A.; Johns, R. A. J. Pharmacol. Exp. Ther. 1996, 276 (1), 30-3.
- (13) Jones, D. P.; Mason, H. S. J. Biol. Chem. 1978, 253 (14), 4874-80.
- (14) Sono, M. Biochemistry 1989, 28 (13), 5400-7.
- (15) Carter, S. K.; Crooke, S. T. Mitomycin C: current status and new developments; Academic Press: New York, 1979.
- (16) Kumar, S.; Malachowski, W. P.; DuHadaway, J. B.; LaLonde, J. M.; Carroll, P. J.; Jaller, D.; Metz, R.; Prendergast, G. C.; Muller, A. J. *J. Med. Chem.* **2008**, *51* (6), 1706–18.
- (17) Carr, G.; Chung, M. K.; Mauk, A. G.; Andersen, R. J. J. Med. Chem. 2008, 51 (9), 2634–7.
- (18) Volgraf, M.; Lumb, J. P.; Brastianos, H. C.; Carr, G.; Chung, M. K.; Munzel, M.; Mauk, A. G.; Andersen, R. J.; Trauner, D. *Nat. Chem. Biol.* **2008**, *4* (9), 535–7.
- (19) Cady, S. G.; Sono, M. Arch. Biochem. Biophys. **1991**, 291 (2), 326–33. (20) Hou, D. Y.; Muller, A. J.; Sharma, M. D.; DuHadaway, J.; Banerjee, T.;
- (20) Hou, D. L., Mullet, A. J., Shahna, M. D., Duhladaway, J., Baherjee, L., Johnson, M.; Mellor, A. L.; Prendergast, G. C.; Munn, D. H. *Cancer Res.* 2007, 67 (2), 792–801.
 (20) Hou, J. C. L. D. G. L. D. G. L. D. G. L. D. G. L. D. F. C. L. D. L. D. K. Stranger, and A. B. S. Stranger, and A. S. Stranger, and A. S. S. Stranger, and A. S. Stranger, and S. S. Stranger, and S. S. Stranger, and S. Stranger, and S. S. Stranger, and S
- (21) Chauhan, N.; Thackray, S. J.; Rafice, S. A.; Eaton, G.; Lee, M.; Efimov, I.; Basran, J.; Jenkins, P. R.; Mowat, C. G.; Chapman, S. K.; Raven, E. L. J. Am. Chem. Soc. 2009, 131 (12), 4186–7.
- (22) Sugimoto, H.; Oda, S.; Otsuki, T.; Hino, T.; Yoshida, T.; Shiro, Y. Proc. Natl. Acad. Sci. U.S.A. 2006, 103 (8), 2611–6.
- JA9029768